

## **Growth of Ca<sub>2</sub>MnO<sub>4</sub> Ruddlesden-Popper structured thin films using combinatorial substrate epitaxy**

M. Lacotte, A. David, D. Pravarthana, C. Grygiel, G. S. Rohrer, P. A. Salvador, M. Velazquez, R. de Kloe, and W. Prellier

Citation: *Journal of Applied Physics* **116**, 245303 (2014); doi: 10.1063/1.4905012

View online: <http://dx.doi.org/10.1063/1.4905012>

View Table of Contents: <http://scitation.aip.org/content/aip/journal/jap/116/24?ver=pdfcov>

Published by the [AIP Publishing](#)

---

The advertisement features a blue background with a film strip graphic on the left. The text is in white and orange. The main headline reads 'Not all AFMs are created equal' in orange, followed by 'Asylum Research Cypher™ AFMs' in white, and 'There's no other AFM like Cypher' in orange. Below this is the website 'www.AsylumResearch.com/NoOtherAFMLikeIt' in white. In the bottom right corner is the Oxford Instruments logo, which consists of the word 'OXFORD' above 'INSTRUMENTS' inside a square frame, with the tagline 'The Business of Science®' below it.

**Not all AFMs are created equal**

**Asylum Research Cypher™ AFMs**

**There's no other AFM like Cypher**

[www.AsylumResearch.com/NoOtherAFMLikeIt](http://www.AsylumResearch.com/NoOtherAFMLikeIt)

**OXFORD**  
INSTRUMENTS  
*The Business of Science®*

# Growth of $\text{Ca}_2\text{MnO}_4$ Ruddlesden-Popper structured thin films using combinatorial substrate epitaxy

M. Lacotte,<sup>1</sup> A. David,<sup>1</sup> D. Pravarthana,<sup>1</sup> C. Grygiel,<sup>2</sup> G. S. Rohrer,<sup>3</sup> P. A. Salvador,<sup>3</sup> M. Velazquez,<sup>4</sup> R. de Kloe,<sup>5</sup> and W. Prellier<sup>1,a)</sup>

<sup>1</sup>Laboratoire CRISMAT, CNRS UMR 6508, ENSICAEN, Université de Basse-Normandie, 6 Bd Maréchal Juin, F-14050 Caen Cedex 4, France

<sup>2</sup>Laboratoire CIMAP, CEA, CNRS UMR 6252, ENSICAEN, Université de Basse-Normandie, 6 Bd Maréchal Juin, F-14050 Caen Cedex 4, France

<sup>3</sup>Department of Materials Science and Engineering, Carnegie Mellon University, 5000 Forbes Ave., Pittsburgh, Pennsylvania 15213, USA

<sup>4</sup>CNRS, Université de Bordeaux, ICMCB, UPR 9048, F-33600 Pessac, France

<sup>5</sup>AMETEK B.V., EDAX Application Laboratory, Tilburg, The Netherlands

(Received 4 November 2014; accepted 11 December 2014; published online 23 December 2014)

The local epitaxial growth of pulsed laser deposited  $\text{Ca}_2\text{MnO}_4$  films on polycrystalline spark plasma sintered  $\text{Sr}_2\text{TiO}_4$  substrates was investigated to determine phase formation and preferred epitaxial orientation relationships (*ORs*) for isostructural Ruddlesden-Popper (RP) heteroepitaxy, further developing the high-throughput synthetic approach called Combinatorial Substrate Epitaxy (CSE). Both grazing incidence X-ray diffraction and electron backscatter diffraction patterns of the film and substrate were indexable as single-phase RP-structured compounds. The optimal growth temperature (between 650 °C and 800 °C) was found to be 750 °C using the maximum value of the average image quality of the backscattered diffraction patterns. Films grew in a grain-over-grain pattern such that each  $\text{Ca}_2\text{MnO}_4$  grain had a single *OR* with the  $\text{Sr}_2\text{TiO}_4$  grain on which it grew. Three primary *ORs* described 47 out of 49 grain pairs that covered nearly all of RP orientation space. The first *OR*, found for 20 of the 49, was the expected RP unit-cell over RP unit-cell *OR*, expressed as  $[100][001]_{\text{film}}||[100][001]_{\text{sub}}$ . The other two *ORs* were essentially rotated from the first by 90°, with one (observed for 17 of 49 pairs) being rotated about the  $[100]$  and the other (observed for 10 of 49 pairs) being rotated about the  $[110]$  (and not exactly by 90°). These results indicate that only a small number of *ORs* are needed to describe isostructural RP heteroepitaxy and further demonstrate the potential of CSE in the design and growth of a wide range of complex functional oxides. © 2014 AIP Publishing LLC. [<http://dx.doi.org/10.1063/1.4905012>]

## I. INTRODUCTION

The Ruddlesden-Popper (RP) type phases are perovskite-like layered oxides of the general formula  $A_{n+1}M_n\text{O}_{3n+1}$  (or  $\text{AO}(\text{AMO}_3)_n$ ),<sup>1–3</sup> where *A* is typically a rare earth or alkaline earth ion and *M* is a transition or post-transition metal ion. The perovskite-like end-members of this family are the  $n = \infty$   $\text{AMO}_3$  perovskites and the  $n = 1$   $\text{A}_2\text{MO}_4$  RP phases (a structure also known the  $\text{K}_2\text{NiF}_4$  structure<sup>1,3</sup>). The structure of the RP family is described by the stacking of *n* consecutive perovskite layers,  $\text{AMO}_3$ , alternating with a single AO rock-salt layer along the crystallographic *c*-direction. A schematic of the RP structure is shown in Figure 1 for the  $n = 1$  end-member. The reduced dimensionality of the *M*-O bonding plays an important role in the properties of RP phases, but the large unit cells and anisotropic nature of the RP crystal structures make synthesis of high-quality crystals, powders, and films more challenging than that of the related perovskites.

RP phases are known to exhibit a range of fascinating functional electronic properties, which can be tuned by changing the nature, size, and valence state of the *A* and *M* cations, non-stoichiometry and chemical order on cation and anion

lattices, as well as through tailoring the dimensionality of the structure, i.e., the *n* value.<sup>3–24</sup> For example, the layered cuprate RP phases played important roles in the discovery of high-temperature superconductivity,<sup>3–6</sup> and the manganese-based RP family displays insulator-to-metal and ferromagnetic transformations at low temperatures, yielding colossal magnetoresistive properties.<sup>3,10,13–15,22–24</sup> The structural, transport, and magnetic properties of nickelate RP compounds have also been extensively investigated,<sup>12,25</sup> being of recent interest as cathode materials in solid oxide fuel cells (SOFCs).<sup>17</sup> Other exotic properties include p-wave superconductivity in the RP ruthenate  $\text{Sr}_2\text{RuO}_4$ ,<sup>16</sup> novel insulating electronic ground states from spin-orbit-coupling in  $\text{Sr}_2\text{IrO}_4$ ,<sup>20,26</sup> and non-centrosymmetry in  $\text{Ca}_2\text{IrO}_4$ .<sup>21</sup> Even the first discovered RP family remains important, as  $\text{Sr}_2\text{TiO}_4$  ( $n = 1$ ) presents advantageous properties for applications in tunable devices and as an alternative gate oxide in metal-oxide semiconductor field-effect transistors (MOSFETs).<sup>27</sup>  $\text{Sr}_4\text{Ti}_3\text{O}_{10}$  ( $n = 3$ ) has been reported for water-splitting photocatalytic activity without any co-catalyst,<sup>28</sup> and novel ordered-structures in the titanate RPs are predicted to be interesting ferroelectrics.<sup>19</sup> The discovery of the  $\text{Ca}_{n+1}\text{Mn}_n\text{O}_{3n+1}$  series followed quickly after the initial description of  $\text{Sr}_{n+1}\text{Ti}_n\text{O}_{3n+1}$ .<sup>1,2,29</sup> These Mn-based materials are of interest for various electronic phenomena, such as

<sup>a)</sup>wilfrid.prellier@ensicaen.fr

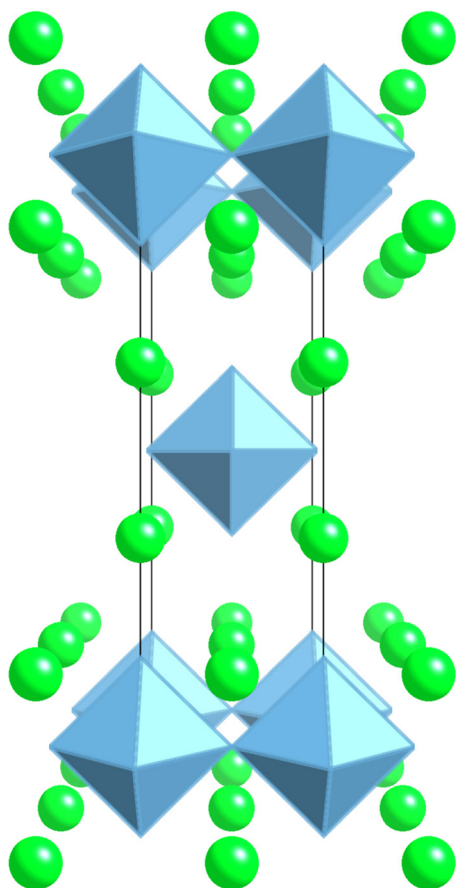


FIG. 1. Schematic of the tetragonal  $A_2MO_4$  Ruddlesden-Popper (or  $K_2NiF_4$ ) structure. The perspective view is looking along the  $a$ -axis, with the  $b$ -axis ( $c$ -axis) along the horizontal (vertical). The spheres represent the  $A$  cations. The  $M$  cations (oxygen) sit at the center (vertices) of the octahedra, which share corners in the  $a$ - $b$  plane but are not connected along the  $c$ -axis.

insulator-to-metal transitions, charge-ordering, and colossal magnetoresistance.<sup>30</sup>

RP phases are typically synthesized through conventional powder/ceramic synthesis. For practical electronic applications, to manipulate properties using substrate-induced strains, or to generate metastable compositions in the RP structure, thin films are of interest. Of course, the substrate choice is of utmost importance in designing and preparing high-quality films. Because of the good crystallographic relationship with the (001) RP plane and their widespread commercial availability, (100)-oriented perovskite single-crystals have been used as substrates for epitaxial growth of RP thin films ( $n = 1$  or 2 usually). However, this choice forces the growth direction to be the [001], which is kinetically more difficult to grow (i.e., requires stricter growth conditions to yield high quality films) and leads to out-of-phase boundaries owing to the existence of energetically degenerate nucleation events.<sup>31</sup> Also, the growth of higher order members ( $n > 2$ ) usually requires advanced layer-by-layer deposition techniques, including reactive-molecular-beam-epitaxy (MBE) or Laser-MBE, and RHEED monitoring.<sup>32–34</sup> These factors render the epitaxial RP growth more challenging than the simpler and more well-studied perovskite, and also restricts our understanding of growth largely to the [001] orientation.

For achieving the best epitaxy, one would ideally use an isostructural substrate, but the choice of commercial single-crystal RP substrates is extremely limited, including  $NdCaAlO_4$  ( $a = 3.685$  Å,  $c = 12.12$  Å),  $SrLaAlO_4$  ( $a = 3.76$  Å,  $c = 12.63$  Å), and  $SrLaGaO_4$  ( $a = 3.84$  Å,  $c = 12.68$  Å). These crystals have small  $a$  lattice constants compared to many of the  $A_2MO_4$  phases described above. Crystal growth on these substrates is often still carried out along the [001],<sup>35–37</sup> but several studies have explored  $a$ -axis growth on the commercially available (100) face, and these films generally grow with high quality, likely owing to the lower kinetic barriers to growth and fewer degenerate nucleation events.<sup>38–40</sup> To understand better RP film growth and to exploit fully the potential of thin film design, it is of interest to broaden the types of RP substrates on which one can grow.

Rather than depositing on commercial single crystals, we have been developing a new strategy that uses well-characterized polycrystalline samples where each grain of the substrate can be viewed as a single crystal of a particular orientation.<sup>41–45</sup> We call this approach Combinatorial Substrate Epitaxy (CSE), in which a very large number of crystallographic orientations are present at the surface of the sample, and one can make statistical analyses of the growth modes across epitaxial orientation space. We have previously demonstrated with CSE, using typical growth conditions to obtain epitaxial films on single crystals, that: (1) grain-over-grain epitaxy occurs for most orientations of substrate grains, (2) only a small number of orientation relationships, ORs, exist for simple structures, (3) similar results are obtained for the growth of heterostructured complex oxide films using CSE as are obtained for films grown on single crystals, and (4) metastable and new complex oxides can be stabilized using CSE. The purpose of this work is to determine if similar observations can be made for RP phases using CSE on isostructural RP substrates, ultimately to improve our understanding of RP film growth.

Here, we use CSE to investigate the growth of the  $Ca_2MnO_4$  ( $a = 3.668$  Å,  $c = 12.050$  Å) thin films on  $Sr_2TiO_4$  ( $a = 3.884$  Å,  $c = 12.600$  Å).<sup>1,30,46</sup> We focus on  $n = 1$  end-members,  $Ca_2MnO_4$ , of the RP oxides, since they are the simplest from a structural perspective. Indeed, layered perovskites are more difficult to synthesize in single-crystalline form than cubic perovskites, and they also exhibit various phenomena including high- $T_C$  or magnetoresistance. The  $Ca_2MnO_4$  has not been investigated so far, and is thus of potential interest.  $Ca_2MnO_4$  thin films were grown using standard pulsed laser deposition (PLD) similarly to previous works on  $La_{2-x}Sr_xNiO_4$  or  $Sr_{2-x}La_xVO_4$ .<sup>40,47</sup> We chose  $Sr_2TiO_4$  as a substrate material because it is the archetypical  $n = 1$  RP oxide, but single crystals cannot be purchased. Instead, we prepared in-house spark plasma sintered polycrystalline substrates. Structural analyses were made using electron backscatter diffraction (EBSD) and grazing-incidence X-ray diffraction (GIXRD). It will be shown that most of the film grew in a grain-over-grain fashion, regardless of the substrate surface orientation, and that most film grains adopted one of only three preferred epitaxial orientations.



## II. EXPERIMENTAL

Commercial powders of  $\text{SrCO}_3$  and  $\text{TiO}_2$  (Cerac, with 99.5% and 99.9% purity, respectively) were used as precursors to synthesize polycrystalline  $\text{Sr}_2\text{TiO}_4$  samples. The precursors were weighed in stoichiometric proportions, were attrition milled for  $\sim 1$  h to reduce and homogenize the grain sizes, and then were calcined 1 h at  $1200^\circ\text{C}$ . The calcined powders were ground in an agate mortar and then loaded between the two punches of a cylindrical graphite die, and protected from external pollution by graphite papers along the die walls. The die was then inserted into a spark plasma sintering (SPS) apparatus (Struers Tegra Force-5) to sinter the powder during 20 min at  $1100^\circ\text{C}$ , under a uniaxial load of 50 MPa. Using this process, larger densities can be obtained in shorter times than using conventional sintering routes.<sup>48,49</sup>

A highly dense pellet of 20 mm diameter was obtained and the structural quality of  $\text{Sr}_2\text{TiO}_4$  was confirmed using X-ray diffraction (not shown). Substrates of  $\sim 5 \times 2 \times 2 \text{ mm}^3$  were cut from the SPS pellets, where the surface of each substrate was taken as the face parallel to the applied pressure. These substrates were meticulously polished for EBSD characterization and film growth, which require flat surfaces of high crystalline quality (in EBSD, only the first tenths of nanometers under the surface are probed<sup>50,51</sup> and the nucleation events are sensitive to the near surface roughness and structure). First, several mechanical polishing steps were performed using SiC papers of different grain sizes, down to  $10 \mu\text{m}$ , to obtain surfaces without any polishing marks. Second, the samples were polished automatically using diamond liquid pastes of grain sizes  $3 \mu\text{m}$  and then  $1 \mu\text{m}$ , to get mirror-like surfaces.  $\text{Ca}_2\text{MnO}_4$  targets were sintered using classical solid state routes.

Thin films (150 nm) were deposited by PLD on the surfaces of the as-prepared  $\text{Sr}_2\text{TiO}_4$  substrates. Depositions were carried out using an  $\text{O}_2$  pressure of  $10^{-3}$  mbar, a laser repetition rate of 2 Hz, and a target-to-substrate distance of 50 mm, leading to a deposition rate around  $0.1 \text{ \AA/pulse}$ . For any given deposition, the substrate temperature was fixed at a value between 650 and  $800^\circ\text{C}$ . The typical thickness of the films was about 30 nm. The composition was verified using energy-dispersive X-ray spectroscopy (EDS).

The substrates and the films were all analyzed by EBSD, using the commercial Orientation Imaging Microscopy (OIM<sup>TM</sup> from EDAX-AMETEK, Inc.) software (v.6.2). This is a recently developed technique based on the acquisition of electron diffraction patterns from a sample placed in a scanning electron microscope (SEM).<sup>52</sup> The sample is tilted to an angle of  $70^\circ$  from the horizontal to maximize the backscattering of electrons.<sup>53</sup> EBSD patterns are formed during a two-stage diffraction process.<sup>54</sup> Briefly, the incident electrons that enter the sample are first scattered inelastically. During this step, the shape/dimensions of the interaction volume are determined. In the second stage of pattern formation, the inelastically scattered electrons travel towards the surface of the tilted sample and interact elastically with the crystal lattice planes. The result is that the bands observed in the diffraction patterns are the trace of a given family of planes, and the intersections of the different bands represent

the relevant zone axes in the volume analyzed. As such, each Kikuchi (EBSD) pattern corresponds to a particular crystallographic orientation.<sup>52</sup>

Using the crystallographic parameters of the phase, the software compares the experimental positions of the Kikuchi bands with the calculated positions of these bands.<sup>55</sup> Several parameters are used to describe the EBSD data, including the Image Quality (IQ), which is a quantitative measure of the intensity of the Hough peaks and describes the crystal quality from the volume that contributes to a measured EBSD pattern,<sup>56</sup> and the Confidence Index (CI), which measures the difference between the best and second best indexing solutions based on a voting algorithm (and ranges from 0 to 1). To compare the patterns of the film and the substrate, typical conditions were used: an SEM voltage of 20 kV, an aperture of  $120 \mu\text{m}$ , and a working distance of 15 mm. By scanning the surface of the sample with a beam step size of  $0.3 \mu\text{m}$ , using a hexagonal grid, many Kikuchi patterns are registered.<sup>55</sup> The software assigns a color pixel for each orientation, and inverse pole figure (IPF) maps of the surface of the substrate and the film are thus recorded. To “clean” the data, points with a confidence index below 0.15 or misorientation angle greater than  $2^\circ$  from the average of the grain were removed, and colored black.

GXRD scans were collected with a grazing incident angle ranging from  $0.3^\circ$  to  $1^\circ$ , to reduce the substrate contribution to X-ray patterns. A Bruker D8 Discover diffractometer was used in  $\theta$ - $\theta$  geometry with a Gobel mirror providing a parallel X-ray beam for a better control of the incident angle. A Cu anode with a  $\text{K}\alpha_1$  line at  $1.54056 \text{ \AA}$  and a  $\text{K}\alpha_2$  line at  $1.54439 \text{ \AA}$  was used.

## III. RESULTS AND DISCUSSION

The EBSD patterns from both the films and substrates generally exhibited strong contrast and were well indexed by the automated software. Fig. 2 displays the average IQ

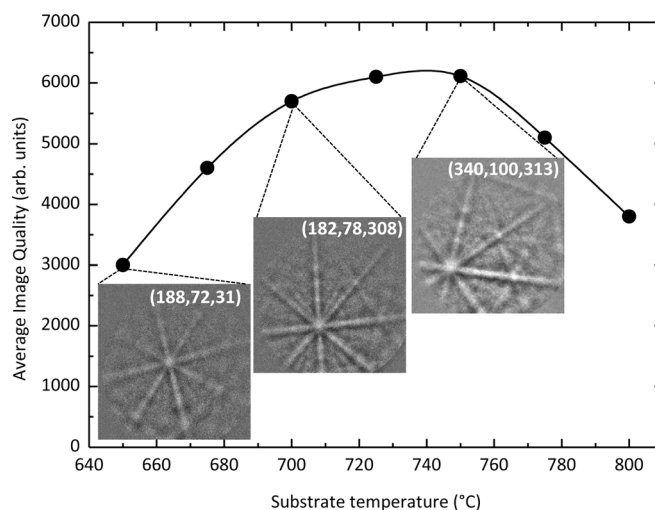


FIG. 2. Graphical representation of the growth optimization methodology used for  $\text{Ca}_2\text{MnO}_4$  films deposited on  $\text{Sr}_2\text{TiO}_4$  polycrystalline substrates. The average image quality (IQ) measured over many grains is plotted as a function of the deposition temperature. Example EBSD (Kikuchi) patterns are also given, and the automated determination of crystallographic orientations is given in Euler angles ( $\phi_1$ ,  $\Phi$ ,  $\phi_2$ ). Units are degrees.

measured over many film grains (of different orientations) deposited at a given temperature, for temperatures ranging from 650 to 800 °C. Representative EBSD patterns from 30 nm thick films are given as insets for depositions carried out at 650, 700, and 750 °C, which illustrate the evolution of the sharpness and brightness of bands. A maximum in the average IQ value occurs between 725 and 750 °C (a maximum was also observed in the average CI for the films deposited at 750 °C). Since these quantitative metrics of crystalline quality maximized for films deposited at 750 °C, this temperature was used for the remainder of the  $\text{Ca}_2\text{MnO}_4$  films studied in the present work. The increase in crystal quality with increasing temperature is likely associated with improved kinetics during nucleation. The decrease in the average IQ (and CI) at higher temperatures is unclear at this time, but may result from changes to a number of temperature dependent parameters, such as a decreased oxygen stoichiometry or increased roughness values, which in turn affect either crystal quality or EBSD image quality.

It should be noted that similar optimizations of growth parameters are routinely made for films deposited on single-crystals, but using X-ray methods (e.g., the full-width-at-half-maximum in the  $\omega$ -scan and/or the  $\theta$ - $2\theta$  scan, the shape of diffraction peaks, the absence of impurities, etc.) In CSE, conventional X-ray methods are not well-suited to the purpose of deposition parameter optimization, because many similarly located peaks arise from both the polycrystalline substrate and film. In the initial investigations of CSE-prepared films, choice of deposition parameters was mainly based on parameters optimized for similar films on single crystals.<sup>41,42,44,45</sup> These results provide a methodology to optimize growth parameters for CSE-prepared films. While we chose to optimize the average IQ as a function of temperature, one could do similar optimizations for any growth parameter, or even for a smaller subset of film orientations. In previous works on RP thin films deposited on single-crystals, the value of the full-width-at-half-maximum of the main diffraction peak was considered to optimize the structural quality.<sup>40,47</sup>

As done previously for CSE grown complex titanate films,<sup>42</sup> one can determine the epitaxial nature of growth by inspection of the EBSD pattern registered from the same grain before and after growth. Fig. 3 shows typical Kikuchi patterns recorded at the surface of the substrate and the film, for two different substrate grains (grains # 5 and # 9 discussed later in Fig. 5). The Kikuchi diagrams corresponding to the substrate (Figs. 3(a) and 3(b)) present sharper and brighter bands than those corresponding to the film measured at the same location (Figs. 3(c) and 3(d)), which is probably a result of relaxation of epitaxial mismatch. Nevertheless, the patterns from the film are of sufficient quality to be characterized using automated software. Furthermore, the bands and zone axes are in the same exact place in the two patterns, showing that the films grew in a unit-cell over unit-cell epitaxial fashion, in these locations. The quality of the films' patterns confirms the smooth flat surfaces, as expected from the average RMS roughness value around 1 nm.

To further confirm the formation of the RP phase, GXRD was performed. Fig. 4 displays patterns recorded,

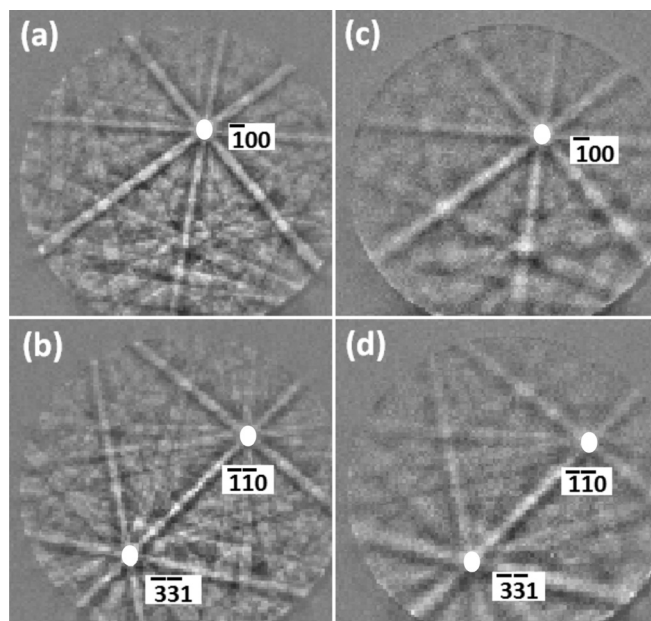


FIG. 3. Kikuchi patterns recorded from two substrate grains, (a) and (b), and two film grains, (c) and (d), grown upon them. (a) and (c) were registered from grain # 5 and (b) and (d) from grain # 9 (discussed later in Fig. 5). The relevant zone axes are indicated in white rectangles. The respective orientations corresponding to (a), (b), (c), and (d) are, in Euler angles, (217,80,297), (231,83,254), (219,84,295), and (234,87,253). Units are degrees.

respectively, at 1° from the surface for the substrate and 1, 0.50, and 0.30° for a 150 nm thick film. For comparison, a theoretical pattern for  $\text{Ca}_2\text{MnO}_4$  was added. Using the material density, the depth probed by the X-ray beam is estimated to be of the order of the film thickness (150 nm) at 0.30° (within the experimental alignment errors). Overall, a dramatic increase of the films' signal-to-background ratio is observed as the angle is reduced. More precisely, small diffraction peaks appear close to  $2\theta = 33^\circ$ ,  $34.6^\circ$ , and  $49.7^\circ$  when the angle is reduced (a small contribution of  $\text{Sr}_2\text{TiO}_4$

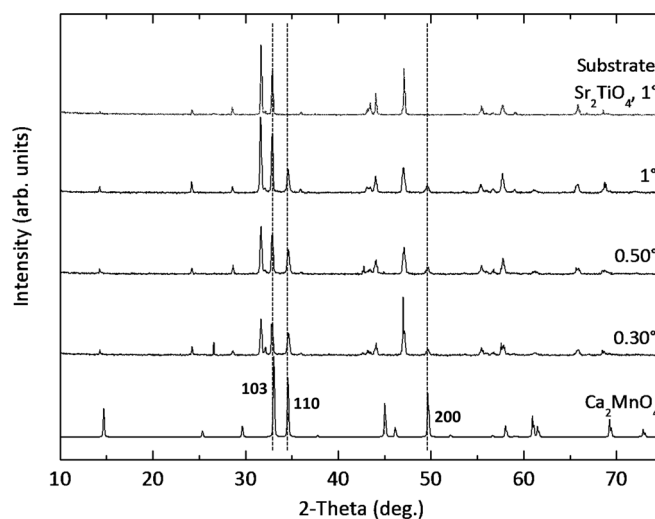


FIG. 4. XRD patterns recorded respectively at 1° from the surface for the  $\text{Sr}_2\text{TiO}_4$  substrate and 1°, 0.50°, and 0.30° for a  $\text{Ca}_2\text{MnO}_4$  film of thickness 150 nm grown upon a  $\text{Sr}_2\text{TiO}_4$  substrate. The theoretical pattern of the  $\text{Ca}_2\text{MnO}_4$  phase is shown for comparison.

substrate is still observed even at  $0.30^\circ$ ). These reflections are assigned to the 103, 110, and 200, respectively, of an  $I4/mmm$  space group (as expected from an RP phase), and are well matched to the expected reflections of the  $\text{Ca}_2\text{MnO}_4$  phase. The analysis therefore suggests the formation of an epilayer of the  $\text{Ca}_2\text{MnO}_4$  phase on top of the  $\text{Sr}_2\text{TiO}_4$  polycrystalline substrate. The corresponding lattice parameters of the film are calculated to be  $a = 3.679 \text{ \AA}$ , and  $c = 12.24 \text{ \AA}$ , which are slightly larger to the expected lattice parameters ( $a = 3.668 \text{ \AA}$ ,  $c = 12.050 \text{ \AA}$ ).<sup>1,30,46</sup>

While confirming phase formation and providing average lattice parameters, the X-ray data do not yield local orientation relationships.

To understand the local growth and epitaxial relationships, IPF maps of the surface of the substrate and the film were recorded from the same area, and are shown in Figs. 5(a) and 5(b). The IPF triangle color code is given in Fig. 5(d). In these maps, points with poor or erroneous indexing (a CI < 0.15) have been colored black; such points correspond to less than 0.6% for the substrate scan and less than 6.8% for the film scan. The average CIs for each map increased from 0.50 to 0.87 for the substrate and 0.18 to 0.67 for the film, respectively, for before and after data cleaning. Enlarged images are shown in Fig. 5(c) for the substrate and Fig. 5(e) for the film. Several grains are clearly recognized, and are identified by numbers 1 to 9. For all of these grains (and most of those observed throughout this investigation), the films grew in a grain-over-grain fashion, where the film has a nearly uniform orientation over the entire substrate grain, terminating at the grain boundary. Moreover, the majority of grains display the same color (orientation) for the substrate and the film grains (# 1, 2, 4, 5, 8, and 9), indicating the crystallographic orientations of these grains are similar to each other. In other words, these grains grew in a unit-cell over unit-cell fashion, even though they have a range of substrate orientations at the surface. Other grains (# 3, 6, and 7) display completely different colors in the IPF maps, indicating that these film grains

have a different orientation from the substrate grains, yet they still grew in a grain-over-grain fashion.

The relationships between the substrate (red circles) and the film (black squares) are summarized in the IPF of Fig. 5(f), for these 9 grains. For the grains # 1, 2, 4, 5, 8, and 9, the symbols corresponding to the same grains on substrate and film are very close in the IPF, and the misorientation angles between substrate and film grains are less than  $7^\circ$ . These values are indicative of nearly perfect unit-cell over unit-cell growth. As an example, grain # 4 is growing with nearly a (100) orientation a similarly oriented substrate grain. The situation is, however, more complex for grains # 3, 6, and 7. Grain # 3 has a (4,1,20) orientation for the substrate, but the same grain is growing near a (211) direction of the film. This change in the epitaxial relationships can be described as a misalignment (tilt) in the grain direction with respect to that of the substrate, and will be discussed at length below. These observations indicate that the growth of these RP films on RP substrates is still influenced by grain orientation, but that only one orientation generally survives to the surface of the film.

We sampled 40 more grain pairs, to have a total of 49 observations of grain-over-grain epitaxial growth, to understand better the preferred epitaxial orientations during RP  $\text{Ca}_2\text{MnO}_4$  on RP  $\text{Sr}_2\text{TiO}_4$  film growth. The ORs between the film/substrate pairs were determined using software and methods described in Zhang *et al.*<sup>41</sup> To illustrate the small number of ORs, we plot in Fig. 6 the angle between the  $[100]_{\text{film}}$  and  $[100]_{\text{sub}}$  (on the horizontal axis) versus the angle between the  $[001]_{\text{film}}$  and  $[001]_{\text{sub}}$  (on the vertical axis). Of these 49 grains, only 4 distinct ORs were observed, represented by clusters of points in the plot (and marked as such). Three primary ORs described 47 out of 49 of the grain pairs, and we will focus on these three in our discussion. OR1 was found for 20 of the 49 pairs (including grains # 1, 2, 4, 5, 8, and 9 discussed above), is represented by a cluster of points near the origin (lower left) in the

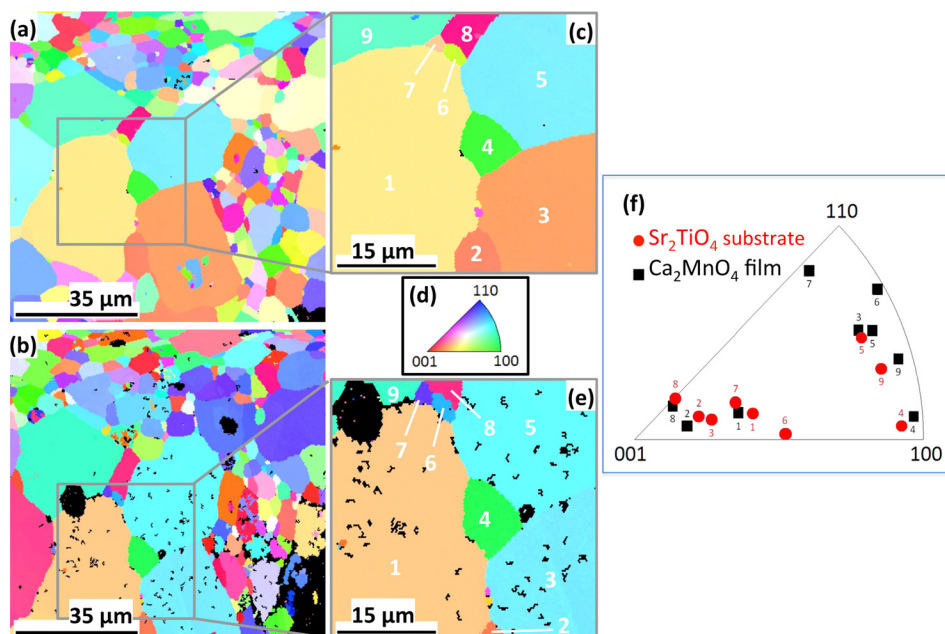


FIG. 5. Inverse pole-figure (orientation) maps of the surface of the substrate (a) and (c), and the film (b) and (e), where (c) and (e) are enlargements of the indicated areas of (a) and (b). The color code for these maps is given as the standard stereographic triangle of a tetragonal crystal in (d). Orientations of the 9 grains in (c) and (e) are plotted in (f) on the standard stereographic triangle, where the numbers correspond to the grain number and a red circle or black square indicates, respectively, a substrate or film grain. The small areas with confidence indices less than the acceptable threshold are colored black in (a)-(e).



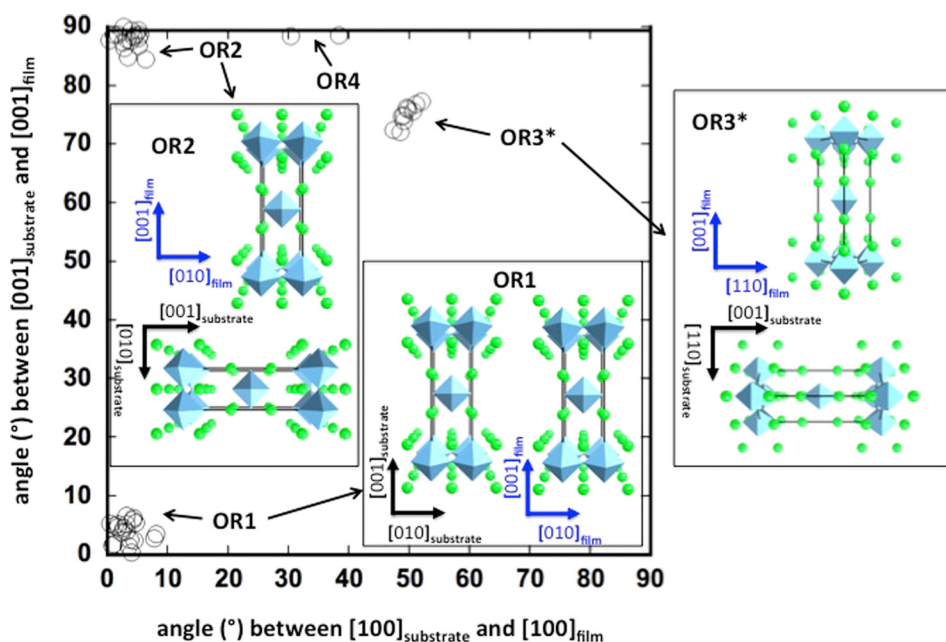


FIG. 6. A plot of the angles of misorientation between the  $[100]$  (horizontal axis) and  $[001]$  (vertical axis) for the film and substrate for the 49 grain pairs plotted in Fig. 7. The insets provide the approximate relationship between the unit cells for ORs 1, 2, and 3. Note that OR 3 is depicted as a  $90^\circ$  rotation about  $[110]$ , which is an oversimplified description (see text). Axes for film and substrate are in black and blue, respectively.

figure, can be expressed as  $[100][001]_{film}||[100][001]_{sub}$ , and is described as the RP unit-cell over RP unit-cell OR. The slight offset from perfect alignment and the variation about the mean are related to relaxations of epitaxial mismatch, as observed elsewhere in CSE grown films.<sup>41,43</sup> The inset marked OR1 indicates the basic crystallographic alignment of the two RP cells for this OR. This may be considered the trivial OR case, as the two unit cells align with one another and, therefore, should have a low-energy (coherent) interface, regardless of the orientation. That this relationship is not observed everywhere is more surprising than the fact that this OR is the most frequently observed OR, at about 40% of the population. When RP phases are grown on RP single crystals, this is the most frequently discussed OR.<sup>31</sup>

OR2 was found for 17 of the 49 pairs (including grain # 3 discussed above), is represented by a cluster of points near the upper left corner of the figure, can be expressed as  $[100][001]_{film}||[100][010]_{sub}$ , and is described as being rotated from OR1 by  $90^\circ$  about the  $[100]$ . The inset marked OR2 indicates the basic crystallographic alignment of the two RP cells for this OR (not accounting for the slight rotations related to epitaxial mismatch). It should be noted that common defects in  $[001]$ -oriented RP films (grown on  $(001)$  RP or perovskite single crystal surfaces) are misoriented regions where the  $c$ -axis is rotated by  $90^\circ$  about the  $\langle 100 \rangle$  axes of the RP structure.<sup>31</sup> While this defect structure is often considered to arise as a result of unfavorable kinetic conditions, which do not allow for the  $c$ -axis to align out of plane, its widespread observation in thin films is an indication that such interfacial planes are of relatively low energy. Therefore, it is not surprising to find OR2 as the second most frequently observed OR, and just slightly less frequent than OR1. Presumably, this orientation is similarly related to kinetic growth factors that inhibit perfect growth of OR1, while also providing a relatively low energy interface. More work is required to understand whether this OR initiates at

the substrate interface directly, or occurs during growth, and they preferentially coarsen until the whole grain has a singular OR. At this thickness and in these specific growth conditions, slightly less about 35% of grains exhibit this OR over the entire grain. Overall, about three quarters of all grains exhibited OR1 or OR2.

OR3 was found for 10 of the 49 pairs (including grains # 3 and 6 discussed above), is represented by a cluster of points near the upper central region of the figure, and is slightly more complicated to describe than the other ORs. If the cluster of points had been located exactly at  $45^\circ$  on the  $x$  axis and  $90^\circ$  on the  $y$  axis, then this OR could be expressed as  $[001][110]_{film}||[110][001]_{sub}$  and could be described as being rotated from OR1 by  $90^\circ$  about the  $[110]$ . The inset marked OR3 indicates the basic crystallographic alignment of the two RP cells for this approximate OR. However, the cluster of points is not quite at  $90^\circ$  on the  $y$ -axis. The actual alignment is about  $16^\circ$  away from this  $90^\circ$  rotation, and can be expressed as  $[223][991]_{film}||[110][001]_{sub}$ . The angle between the  $[223]$  and  $[991]$  is about  $89^\circ$ , so they can align reasonably well with the  $90^\circ$  difference between the  $[110]$  and  $[001]$ . The difference between the approximate relationship, shown as the inset OR3, and the real OR is simply a  $16^\circ$  rotation about the  $[110]$ . The cause of this “extra” rotation is unclear at this time, and represents an interesting observation made possible by CSE. To our knowledge, this OR represents a new OR for RP film growth and about 20% of all grains have this OR over the entire grain. As for OR2, more work is required to understand whether OR3 initiates at the substrate interface directly, or occurs during growth, and they preferentially coarsen until the whole grain has a singular OR. Overall, about 95% of all grains exhibited OR1, OR2, or OR3.

An obvious question to ask is: are the different ORs described above correlated in some way to the substrate surface normal of the grain on which it grew? Rather than plotting the pairs of substrates on the same stereographic

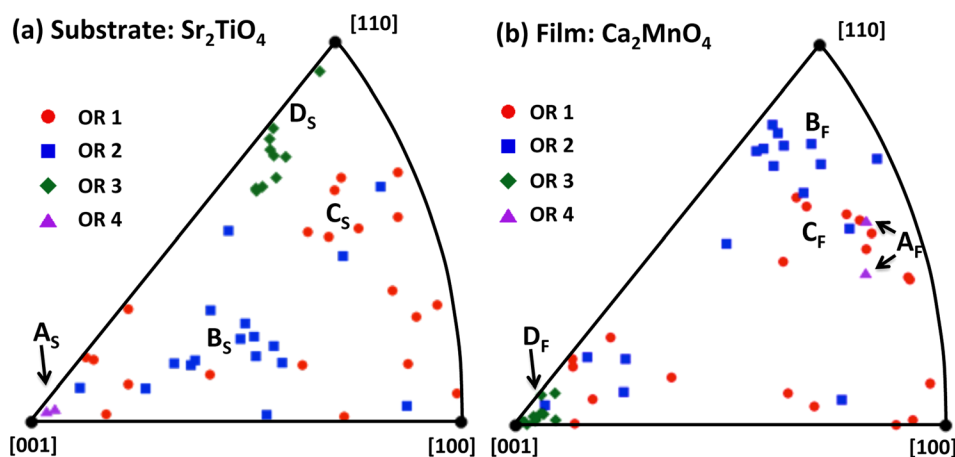


FIG. 7. Orientations of 49 grain pairs that were used to determine the epitaxial ORs, plotted using the standard stereographic triangle. In both (a) and (b), the marker type indicates what OR was found for the grain with that orientation, as indicated in the key. (a) Marker locations indicate the orientation of the substrate,  $\text{Sr}_2\text{TiO}_4$ . (b) Marker locations indicate the orientation of the film,  $\text{Ca}_2\text{MnO}_4$ .

triangle, as done in Fig. 5(f), we plot the four ORs using different symbols on two stereographic triangles in Fig. 7. In Fig. 7(a), the ORs are plotted with respect to the substrate surface normal, while in Fig. 7(b) they are plotted with respect to the film surface normal. Largely speaking, OR1 and OR2 are distributed throughout the stereogram, with a slight increase in OR2 in the central region of the triangle. OR3 shows up almost exclusively near the edge of the triangle, near [110] extending in the direction of [001]. For single crystal growth on RP single crystals, the [110] orientations have not been investigated, and this may explain why OR3 has not been reported. OR4 was observed only near the [001] oriented substrate crystals. With only two observations, it is difficult to make any strong conclusion here (the epitaxy of OR4 can be expressed as  $[212]_{\text{film}}||[520]_{\text{film}}||[110][001]_{\text{sub}}$ ), and nearby both OR1 and OR2 are also observed.

It is also interesting to explore how the ORs are distributed with respect to the film surface normal, as shown in Fig. 7(b). As expected, OR1 remains distributed throughout the triangle (with decreased frequencies where OR2 and OR3 had clusters in Fig. 7(b)). OR2 is also distributed throughout, but has a larger cluster of observation near the [110] orientation (related to the cluster in the center of Fig. 6(b)), which have rotations similar to grain 7 in Fig. 5(f). Some of OR1 and OR2 are clustered with an orientation near [001], as are almost all of OR3. What is surprising about this observation is that growth along the [001] is considered kinetically challenging owing to the large repeat period along that direction. For OR3, where the c-axis is nearly in the plane of the substrate surface, the film grows with the c-axis nearly perpendicular to the substrate surface. This observation seems counterintuitive to the accepted kinetic models of RP film growth, where c-axis films are difficult to obtain, and implies that there is more to understand for epitaxial growth on general surfaces of crystals. Perhaps, c-axis films could be obtained more readily on RP orientations near the 110 since the latest being the most thermodynamically stable planes. More work is necessary to unravel the details and origin of this seemingly anomalous OR, as is true also for the nature of OR2.

#### IV. CONCLUSIONS

RP type  $\text{Ca}_2\text{MnO}_4$  thin films were grown by pulsed laser deposition on isostructural  $\text{Sr}_2\text{TiO}_4$  polycrystalline substrates

prepared by SPS. Optimization of the film structural quality as a function of deposition temperature was made by considering the pattern quality from electron backscatter diffraction. Grazing-incidence X-ray diffraction of the film deposited at  $750^\circ\text{C}$  confirmed the RP phase formation. The Kikuchi patterns displayed bright and sharp bands, attesting to the local formation of the RP phases everywhere on the substrate. Inverse Pole Figure maps recorded on the same grain before and after film deposition revealed a perfect grain-over-grain growth for the majority of grains. Three primary ORs described 47 out of 49 grain pairs substrate-film that covered nearly all of RP orientation space. The first OR, found for 20 of the 49, was the expected RP unit-cell over RP unit-cell OR, expressed as  $[100][001]_{\text{film}}||[100][001]_{\text{sub}}$ . The other two ORs were essentially rotated from the first by  $90^\circ$ , with one (observed for 17 of 49 pairs) being rotated about the [100] and the other (observed for 10 of 49 pairs) being rotated about the [110] (and not exactly by  $90^\circ$ ). These results indicate that only a small number of ORs are needed to describe isostructural RP heteroepitaxy and further demonstrate the potential of CSE in the design and growth of a wide range of complex functional oxides.

#### ACKNOWLEDGMENTS

We thank L. Gouleuf and J. Lecourt for technical support. M. Lacotte received her Ph.D. scholarship from the Ministère de l'Enseignement Supérieur et de la Recherche. D. Pravarthana is supported by a Ph.D. fellowship included in the Erasmus Mundus Project IDS-FunMat. Partial support of the French Agence Nationale de la Recherche (ANR), through the program Investissements d'Avenir (ANR-10-LABEX-09-01), LabEx EMC3, and the Interreg IVA MEET project are also acknowledged. We also thank Dr. O. Perez, Dr. I. Canero-Infante, Professor B. Mercey, and B. Raveau for fruitful discussions.

<sup>1</sup>S. Ruddlesden and P. Popper, *Acta Crystallogr.* **10**, 538 (1957).

<sup>2</sup>S. Ruddlesden and P. Popper, *Acta Crystallogr.* **11**, 54 (1958).

<sup>3</sup>I. Sharma and D. Singh, *Bull. Mater. Sci.* **21**, 363 (1998).

<sup>4</sup>A. Dwivedi and A. N. Cormack, *Bull. Mater. Sci.* **14**, 575 (1991).

<sup>5</sup>N. Nguyen, J. Choisnet, M. Hervieu, and B. Raveau, *J. Solid State Chem.* **39**, 120 (1981).

<sup>6</sup>J. G. Bednorz and K. A. Müller, *Z. Phys. B* **64**, 189 (1986).

<sup>7</sup>P. D. Battle, S. K. Bollen, and A. V. Powell, *J. Solid State Chem.* **99**, 267 (1992).



- <sup>8</sup>M. P. Attfield, P. D. Battle, S. K. Bollen, S. H. Kim, A. V. Powell, and M. Workman, *J. Solid State Chem.* **96**, 344 (1992).
- <sup>9</sup>J. Y. Lee, J. S. Swinnea, H. Steinfink, W. M. Reiff, S. Pei, and J. D. Jorgensen, *J. Solid State Chem.* **103**, 1 (1993).
- <sup>10</sup>R. Mahesh, R. Mahendiran, A. K. Raychaudhuri, and C. N. R. Rao, *J. Solid State Chem.* **122**, 448 (1996).
- <sup>11</sup>P. D. Battle, M. A. Green, N. S. Laskey, J. E. Millburn, M. J. R. L. Murphy, S. P. Sullivan, and J. F. Vente, *Chem. Mater.* **9**, 552 (1997).
- <sup>12</sup>M. Greenblatt, *Curr. Opin. Solid State Mater. Sci.* **2**, 174 (1997).
- <sup>13</sup>I. D. Fawcett, J. E. Sunstrom, M. Greenblatt, M. Croft, and K. V. Ramanujachary, *Chem. Mater.* **10**, 3643 (1998).
- <sup>14</sup>P. D. Battle and M. J. Rosseinsky, *Curr. Opin. Solid State Mater. Sci.* **4**, 163 (1999).
- <sup>15</sup>T. Kimura and Y. Tokura, *Annu. Rev. Mater. Sci.* **30**, 451 (2000).
- <sup>16</sup>F. Lichtenberg, *Prog. Solid State Chem.* **30**, 103 (2002).
- <sup>17</sup>G. Amow and S. J. Skinner, *J. Solid State Electrochem.* **10**, 538 (2006).
- <sup>18</sup>S.-W. Cheong, *Nature Mater.* **6**, 927 (2007).
- <sup>19</sup>S. Nakhmanson, *Phys. Rev. B* **78**, 064107 (2008).
- <sup>20</sup>H. Zhang, K. Haule, and D. Vanderbilt, *Phys. Rev. Lett.* **111**, 246402 (2013).
- <sup>21</sup>P. V. Balachandran, D. Puggioni, and J. M. Rondinelli, *Inorg. Chem.* **53**, 336 (2014).
- <sup>22</sup>C. D. Ling, J. E. Millburn, J. F. Mitchell, D. N. Argyriou, J. Linton, and J. H. N. Bordallo, *Phys. Rev. B* **62**, 15096 (2000).
- <sup>23</sup>D. N. Argyriou, J. F. Mitchell, P. G. Radaelli, H. N. Bordallo, D. E. Cox, M. Medarde, and J. D. Jorgensen, *Phys. Rev. B* **59**, 8695 (1999).
- <sup>24</sup>J. F. Mitchell, D. N. Argyriou, J. D. Jorgensen, D. G. Hinks, C. D. Potter, and S. D. Bader, *Phys. Rev. B* **55**, 63 (1997).
- <sup>25</sup>Z. Kakol, J. Spalek, and J. M. Honig, *J. Solid State Chem.* **79**, 288 (1989).
- <sup>26</sup>L. Miao, H. Xu, and Z. Q. Mao, *Phys. Rev. B* **89**, 035109 (2014).
- <sup>27</sup>J. H. Haeni, C. D. Theis, D. G. Schlom, W. Tian, X. Q. Pan, H. Chang, I. Takeuchi, and X.-D. Xiang, *Appl. Phys. Lett.* **78**, 3292 (2001).
- <sup>28</sup>Y.-G. Ko and W. Y. Lee, *Catal. Lett.* **83**, 157 (2002).
- <sup>29</sup>M. Velazquez, C. Haut, B. Hennion, and A. Revcolevshi, *J. Cryst. Growth* **220**, 480 (2000).
- <sup>30</sup>C. Autret, C. Martin, M. Hervieu, R. Retoux, B. Raveau, G. Andre, and F. Boure, *J. Solid State Chem.* **177**, 2044 (2004).
- <sup>31</sup>M. A. Zurbuchen, W. Tian, X. Q. Pan, D. Fong, S. K. Streiffer, M. E. Hawley, J. Lettieri, Y. Jia, G. Asayama, S. J. Fulk, D. J. Comstock, S. Knapp, A. H. Carim, and D. G. Schlom, *J. Mater. Res.* **22**, 1439 (2007).
- <sup>32</sup>W. Tian, J. H. Haeni, D. G. Schlom, E. Hutchinson, B. L. Sheu, M. M. Rosario, P. Schiffer, Y. Liu, M. A. Zurbuchen, and X. Q. Pan, *Appl. Phys. Lett.* **90**, 022507 (2007).
- <sup>33</sup>R. G. Palgrave, P. Borisov, M. S. Dyer, S. R. C. McMitchell, G. R. Darling, J. B. Claridge, M. Batuk, H. Tan, H. Tian, J. Verbeeck, J. Hadermann, and M. J. Rosseinsky, *J. Am. Chem. Soc.* **134**, 7700 (2012).
- <sup>34</sup>L. Yan, H. Niu, C. A. Bridges, P. A. Marshall, J. Hadermann, G. van Tendeloo, P. R. Chalker, and M. J. Rosseinsky, *Angew. Chem.* **46**, 4539 (2007).
- <sup>35</sup>J.-P. Locquet, J. Perret, J. Fompeyrine, E. Machler, J. W. Seo, and G. Van Tendeloo, *Nature* **394**, 453 (1998).
- <sup>36</sup>W. Si and X. X. Xi, *Appl. Phys. Lett.* **78**, 240 (2001).
- <sup>37</sup>I. Bozovic, G. Logvenov, I. Belca, B. Narimbetov, and I. Sveklo, *Phys. Rev. Lett.* **89**, 107001 (2002).
- <sup>38</sup>S. Madhavan, D. G. Schlom, A. Dabkowski, H. A. Dabkowska, and Y. Liu, *Appl. Phys. Lett.* **68**, 559 (1996).
- <sup>39</sup>W. Si and X. X. Xi, *Appl. Phys. Lett.* **78**, 240 (2001).
- <sup>40</sup>S. Shinomori, M. Kawasaki, and Y. Tokura, *Appl. Phys. Lett.* **80**, 574 (2002).
- <sup>41</sup>Y. Zhang, A. M. Schultz, L. Li, H. Chien, P. A. Salvador, and G. S. Rohrer, *Acta Mater.* **60**, 6486 (2012).
- <sup>42</sup>S. Havelia, S. Wang, K. R. Balasubramaniam, A. M. Schultz, G. S. Rohrer, and P. A. Salvador, *CrystEngComm* **15**, 5434 (2013).
- <sup>43</sup>A. M. Schultz, Y. Zhu, S. A. Bojarski, G. S. Rohrer, and P. A. Salvador, *Thin Solid Films* **548**, 220 (2013).
- <sup>44</sup>D. Pravarthana, O. I. Lebedev, S. Hebert, D. Chateigner, P. A. Salvador, and W. Prellier, *Appl. Phys. Lett.* **103**, 143123 (2013).
- <sup>45</sup>D. Pravarthana, M. Trassin, J. H. Chu, M. Lacotte, A. David, R. Ramesh, P. A. Salvador, and W. Prellier, *Appl. Phys. Lett.* **104**, 082914 (2014).
- <sup>46</sup>M. E. Leonowicz, K. R. Poeppelmeier, and J. M. Longo, *J. Solid State Chem.* **59**, 71–80 (1985).
- <sup>47</sup>J. Matsuno, Y. Okimoto, M. Kawasaki, and Y. Tokura, *Appl. Phys. Lett.* **82**, 194 (2003).
- <sup>48</sup>Z. A. Munir, D. V. Quach, and M. Ohyanagi, *J. Am. Ceram. Soc.* **94**, 1 (2011).
- <sup>49</sup>M. Nygren and Z. Shen, *Solid State Sci.* **5**, 125 (2003).
- <sup>50</sup>V. Randle, *Mater. Charact.* **60**, 913 (2009).
- <sup>51</sup>K. Z. Baba-Kishi, *Scanning* **20**, 117 (1998).
- <sup>52</sup>A. J. Schwartz, M. Kumar, B. L. Adams, and D. P. Field, *Electron Backscatter Diffraction in Materials Science* (Springer, 2009).
- <sup>53</sup>A. Winkelmann and G. Nolze, *Ultramicroscopy* **110**, 190 (2010).
- <sup>54</sup>R. P. Goehner and J. R. Michael, *J. Res. Natl. Inst. Stand. Technol.* **101**, 301 (1996).
- <sup>55</sup>B. L. Adams, S. I. Wright, and K. Kunze, *Metall. Trans. A* **24**, 819 (1993).
- <sup>56</sup>S. I. Wright and M. M. Nowell, *Microsc. Microanal.* **12**, 72 (2006).